

Solution Combustion Synthesis of Nickel Oxide and Reduction to Ni Nano Particles for the Synthesis of Graphitic Carbon

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ABSTRACT

Synthesis of nano carbon and its different forms like CNT, graphite, graphene, DLC is a popular topic of research nowadays due to their applications in various fields such as catalysis, supercapacitors, gas sensors, bio-sensors etc. For the synthesis of graphite carbon; synthesis of Nickel nanoparticles is an important step, which is used as a catalyst. So in the present study our efforts were to synthesize NiO nanoparticles using solution combustion synthesis method and its reduction to Ni nanoparticles. NiO particles were prepared by using urea thermal decomposition technique. Aqueous solution of Ni (NO₃)₂·6H₂O and urea with different molar ratios was heated to 400°C temperature to give NiO particles; which when reduced in H₂ atmosphere at 600°C temperature for 2 hours produced Ni nanoparticles. Percentage yield of Ni nanoparticles was 8-10%. Graphitic carbon is synthesized by CVD at 800°C by using oil as a precursor and Ni nanoparticles as a catalyst. The surface morphology and nature of graphitic carbon obtained was confirmed by SEM and XRD studies respectively.

Keywords: CVD, Solution combustion, Nickel oxide, Nano particles