

Limnological Studies of Pedda Cheruvu Lake in Rajgopalpet, Siddipet District, Telangana State

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ABSTRACT

Article Info	Pure water is the world's first and foremost medicine and lakes are inland water
Volume 8, Issue 1	bodies of water. Limnological studies of peddacheruvu Lake, Rajgopalpet,
Page Number: 212-222	Siddipet district, Telangana state was carried out was over period of two years.
	The physico-chemical characteristics were studied and analyzed during the
Publication Issue :	period of one year. Seasonal variations at three different stations of the lake were
January-February-2021	observed. Sewage drains, clothes washing, bathing of animals, agro-waste with
	pesticides residue and rituals waste drains into the lake may result into the
	change in physico-chemical characteristics of lake water. Some parameters like
	pH, temperature, Dissolved oxygen, Hardness, Nitrate, silicates and Phosphates
Article History	etc. studied throughout year.
Accepted : 08 Feb 2021	Keywords : Limnological study, Physico-chemical, variations, peddacheruvu
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I. INTRODUCTION

Water is known to contain many chemical elements. The interaction of both the physical and chemical properties of water plays a significant role in the composition, distribution and abundance of aquatic ecosystem. Apart from this, it also gives an insight into the relationships between the organism and their environment and can be used in determining water quality, and productivity the lake. The of physicochemical study could also help in understanding the structure and function of a particular water body in relation its habitants. The proper balance of physical, chemical and biological properties of water in lake is an essential ingredient for successful production of aquatic resources.

Water is one of the important sources, to sustain life and has long been suspected of being the source of much human illness. Source of surface water and ground water have become increasingly contaminated due to increased industrial and agricultural activity. The public desires water that is low in hardness total solids, non-corrosive and non-scale-forming.

A fascinating thing about lakes is that they provide their own variety. The task of limnologist is to measure and interpret this variation whether it concerns physical, chemical, biological phenomena, altitude, geology of the catchments area and the depth of water. Increased use of lakes and reservoirs for recreation, fisheries, water storage for irrigation and

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electric power generation purpose as well as for urban shoreline development, has emphasized the need for intensive water quality studies and management. The physical and chemical limnology of a lake is characterized by hydrological impact, autogenic nutrient dynamic and biological aspects. These factors combine with each other determine the water quality and consequently community of the lake.

Rajgopalpet is a village in Nangunoor mandal, siddipet district in Indian state of Telangana. It is the biggest town in Siddipet district and is located in siddipet to Warangal route. Rajgopalpet is located at Latitude 18°09'60"38N, Longitude 78°93'73"93E. It has an average elevation of 675 meters (2215feet) above the mean sea level. The village of Rajgopalpet climate was once fairly equitable with maximum temperature 43°C and minimum temperature 12°C, avg. Annual temperature 30°C, avg. summer temperature 43°C, and avg. winter is 16°C.

II. METHODS AND MATERIAL

The samples for physico-chemical analysis were collected from peddacheruvu lake, Rajgopalpet, in a period of one year. Separate samples were collected to estimate dissolved oxygen in 250ml BOD bottles and dissolved oxygen was fixed in the field by adding Wrinkler's reagent immediately after sample collection. The temperature and pH were measured with the help of thermometer and digital pH meter at collection site The physico-chemical only. characteristics of the lake water like dissolved oxygen, Total alkalinity, Hardness, silicates, Nitrates and Phosphates were determined in every month according to standard procedure from APHA (2005), Trivedi and Goel (1986).

III. RESULTS AND DISCUSSION

Temperature: Water has unique thermal property which combines to minimize temperature changes. It affects on biochemical reactions, population fluctuation of water body as well as physical and chemical characteristics of water. During the present investigation the surface water temperature of lake showed considerable fluctuation. The temperature varies between a December minimum of 22.0°C to a June maximum 25.9°C. The pattern of variation in water temperature was being the lowest in January.

pH:

pH of surface water ranges between 7.8 to 8.2 and indicated alkaline nature of the water. pH concentrations was altered with seasons (Ruth *et.al.*, 2013).

Carbonates:

The occurrence of carbonates in a water body is of immense relevance so much so that most of the physico chemical parameters have a direct bearing with it. In fact it is the prime contributor for maintaining pH of a water body and ends its role is of vital importance (Hegde et.al. 2005). The highest amount of carbonates found to be 51.92 mg/l in May and the lowest value was recorded as 25.54 mg/l in June.

Bicarbonates:

The greater amount of bicarbonates were identified as 240.64 mg/l in August due to the accumulation of organic matter produced by decomposition of vegetation and lowest value is 120.34 mg/l in October due to the inflow of freshwater and dissolution of calcium carbonate ions in the water.

Dissolved Oxygen (D.O.):

Oxygen distribution is key criterion as it is the direct need of most of the organism present in the environment. It also affects the solubility and availability of many nutrients and therefore the productivity of aquatic ecosystem. The dissolved oxygen varied in the range of 8.2 to 11.4 mg/l of observations with an average of 9.77 mg/l. The fluctuation in dissolved oxygen level remained marginal during the period of present investigations.

During summer lowest values of dissolved oxygen may be due to high temperature and low solubility of oxygen in water consequently affecting the BOD. Further, the dissolved oxygen content of water was low in summer because of its enhanced utilization by microorganisms in the decomposition of organic matter (Naik *et.al.*, 2012).

Biochemical Oxygen Demand (BOD):

In the present work, during the period of investigation, BOD values ranged from 4 mg/l to 10 mg/l in the Peddacheruvu lake. BOD is the major criterion used in stream pollution control where organic loading must be restricted to maintain desired dissolved oxygen. In unpolluted water BOD is lower while it is high in polluted water.

Total Hardness:

Total hardness ranged between 260.0 mg /l maximum and a minimum value 146.0mg/l. was during February. The total hardness values varies widely suggested that the concentration of total hardness might be increased due to input of domestic and other sewage water in the lake. According to Jhingran (1991) soft water lakes are generally poorer in regard to their aquatic fauna and flora and usually contain less living matter per unit area than hard water lakes. Although the total mass of organisms is greater in hard water lakes while medium lakes hold a greater variety of living organisms.

Calcium:

Calcium is an important nutrient for aquatic organism and it is commonly present in all water bodies (Ansari and Prakash, 2000). The maximum quantity of calcium was recorded in December as 53.96 mg/l and minimum 36.24 mg/l amount was determined during February.

Magnesium:

Magnesium is essential for chlorophyll bearing plant for photosynthesis and act as a limiting factor for the growth of phytoplankton. The highest estimated value of magnesium was 40.22 mg/l in the month of September and the lowest value was 22.78 mg/l found in August. Lowering of magnesium level minimizes the phytoplankton population (Govindan and Devika., 1991).

Total Dissolved Solids (TDS):

The water always contains some dissolved solids in natural condition. The solids remaining in water after filtration are called total dissolved solids. They may be organic or inorganic but precisely, the dissolved solids are composed mainly of carbonates, bicarbonates, chlorides, sulphates, calcium, phosphate and iron (Trivedy, 1986). The greater TDS value for peddacheruvu lake water was estimated to be 435 mg/l in the month of April and lower was 310 mg/l in the month of August.

Chlorides:

The estimated highest concentration of chloride was 196.98 mg/l in June. The higher concentration of chloride is considered to be an indicator of higher pollution of due to higher organic waste of animal origin (Ramakrishna *et.al.*,2005). Due to dilution of lake water with rain water the concentration of chlorides decreases and found 126.25 mg/l in January.

Sulphates:

Sulphates are found in all natural water, especially those with high salt content. Besides industrial pollution and domestic sewage and biological oxidations of reduced sulphur species also add to sulphate content. The highest amount of sulphates was 38 mg/l during the month of May and lowest was 22 mg/l in the September month during the period of investigation.

Phosphates:

Phosphorus is an essential nutrient for primary producer; hence act as one of the limiting factors in the process of eutrophication and lakes can be aesthetically classified into good, fair and bad on the basis of % phosphates loading. In natural water, phosphorus is present in very small quantities. The main supply of phosphorus only in natural water is from the withering of phosphorus bearing rocks and leaching of the soils of the catchment area by rain. Generally excess of this nutrient through untreated domestic sewage and agricultural runoff lack of phosphorus content of more than 0.20 mg/l are likely to be quite productive. The phosphate-phosphorus was recorded in the range of 0.38 to 0.98 mg/l of observations.

Nitrates:

It is well known fact that the role of the nitrates in biological productivity of aquatic ecosystem. In an aquatic environment, nitrogen is present in combined forms of ammonia, nitrite, nitrate, urea and dissolved organic compounds nitrate ranged between 0.58 to 1.14 mg/l. The marked increase in the nitrates level was observed during months of June probably due to decomposition of macrophytes.

Nitrites:

In the fresh water nitrites are usually in low quantities compare to other nitrogenous forms. They represent an intermediate form in de-nitrification and nitrification reactions in the nitrogen cycle. The major source of nitrites in the water body is biological decomposition. The estimated maximum amount of nitrites was 0.06 mg/l due to the possible inflow of nitrogen rich flood water into the lake. The lowest value was 0.02 mg/l.

Silicates:

The importance of silicates in the production of algal growth is well recognized. The decreased amount of Silicates may occur when there is an abundance of diatoms. It has been considered as an important parameter regulating the diatomic population in fresh water eco systems. In the present investigation the value ranged 1.84 to 2.99 mg/l.

Organic matter:

The organic debris and the substances which are biological in nature generate the oxidizable organic matter. The infusion of organic matter into ponds and lakes is in the form of external sources such as inflowing water from the areas outside the besin. The highest value of organic matter was found to be 1.6 mg/l in September and lowest value was recorded of 0.6 mg/l in August.

Chemical Oxygen Demand (COD):

The estimation of COD along with BOD is helpful in indicating toxic conditions and the presence of non biodegradable substances in the water (Sawyer, McCarty & Parkin, 2003). High amount of COD indicates presence of all forms of organic matter, both biodegradable and non biodegradable and degree of pollution of lake water. The estimated greater amount of COD was 28 mg/l in January and lowest was 14 mg/l in May.

In the present investigation an attempt was made to generate base line information about some physicochemical characteristics and quality of peaddachearuvu lake water in different season. The winter, summer and monsoon seasons shows seasonal fluctuations in various physico-chemical parameters. The water of present lake is utilized for irrigation, washing of cloths, fish culture and rituals. The water parameters indicate that the lake is rich in nutrients.

Station-I

Table 1 - Monthly variation of Physico-chemical parameters of Peddacheruvu Lake water

	Oct- 2018	Nov	Dec	Jan- 2019	Feb	Mar	Apr	May	June	July	Aug	Sep- 19	Avg.
Temp (in ⁰ C)	22.4	22.1	22.0	23.5	24.6	24.9	25.2	25.6	25.9	24.2	23.2	22.9	23.87
рН	7.9	7.9	7.9	7.8	8.0	8.0	8.1	8.2	8.1	8.2	8.1	8.2	8.03
Carbonates	41.24	46.26	26.24	51.61	31.24	46.31	36.21	51.92	25.56	39.34	48.42	42.64	40.58
Bicarbonates	232.34	226.54	286.64	234.22	240.28	228.74	238.54	230.44	278.38	254.38	250.56	230.26	244.27
DO	9.2	10.4	8.2	8.6	10.2	9.2	9.4	9.2	9.6	10.2	11.2	11.4	9.73
BOD	10	8	10	8	6	8	10	8	8	4	4	4	7.33
ТН	170	184	208	202	240	178	186	194	168	172	132	184	184.83
Calcium	40.12	51.28	53.58	48.86	36.24	44.28	44.96	46.92	41.24	42.66	42.28	44.64	44.75
Magnesium	28.84	29.64	26.96	34.69	26.84	30.68	32.26	34.66	26.24	30.66	22.78	34.22	29.87
TDS	340	330	320	320	360	400	410	380	320	340	310	360	349.16
Chlorides	160.54	138.46	190.58	126.25	168.28	120.86	156.82	146.56	190.26	156.56	146.64	138.94	153.39
Sulphates	28	26	36	32	36	30	34	30	34	32	36	24	31.5
Phosphates	0.62	0.88	0.66	0.88	0.66	0.52	0.94	0.44	0.56	0.42	0.76	0.76	0.67
Nitrates	0.66	0.74	0.94	0.86	0.84	0.64	0.66	0.98	1.12	1.08	0.98	0.96	0.87
Silicates	1.86	2.46	1.54	2.24	2.44	2.14	1.76	2.74	2.98	1.64	1.78	2.14	2.14
Organic matter	1.6	0.8	1.4	1.6	0.8	1.2	1.6	1.6	0.6	0.8	0.6	1.6	1.18
COD	20	24	16	28	24	18	16	14	16	18	16	14	18.66
Nitrite	0.04	0.02	0.04	0.04	0.06	0.05	0.05	0.04	0.04	0.06	0.02	0.04	0.04





Tuble 2 monthly valuation of Thyseo chemical parameters of Tedaacheruvu Lake water													
	Oct-	Nov	Dec	Jan-	Feb	Mar	Apr	May	June	July	Aug	Sep-19	Avg.
	2018			2019									
Temp(in ⁰ C)	22.4	22.1	22.0	23.5	24.6	24.9	25.2	25.6	25.9	24.2	23.2	22.9	23.87
pH	7.9	7.8	7.8	7.8	7.9	8.0	8.0	8.1	8.2	8.1	8.2	8.2	8.0
Carbonates	40.21	44.22	28.26	50.60	30.24	46.31	36.22	50.98	25.54	38.98	47.98	42.64	40.18
Bicarbonates	232.34	225.98	286.64	234.20	239.28	228.74	238.56	231.44	277.96	254.38	250.58	231.98	244.34
DO	9.8	10.2	8.2	8.4	9.8	9.2	9.2	9.4	9.2	10.4	11.4	11.2	9.7
BOD	8	8	10	10	6	8	8	10	10	6	6	4	7.83
ТН	175	186	210	210	250	182	190	196	172	182	146	196	191.25
Calcium	44.12	52.26	52.48	49.86	38.28	46.28	46.94	50.22	42.28	44.68	48.32	49.28	47.08
Magnesium	29.86	28.66	28.98	36.88	30.82	32.66	34.36	38.64	28.28	36.22	28.76	38.66	38.73
TDS	360	340	330	320	380	420	430	400	330	360	320	360	362.5
Chlorides	162.52	140.44	180.56	130.25	170.28	140.86	158.82	148.56	196.28	160.52	148.66	142.92	150.72
Sulphates	30	30	34	36	38	38	36	38	32	30	38	22	33.5
Phosphates	0.68	0.78	0.86	0.68	0.60	0.56	0.98	0.58	0.56	0.38	0.66	0.68	0.66
Nitrates	0.64	0.72	0.92	0.82	0.82	0.60	0.58	0.88	1.14	1.08	1.02	0.98	0.85
Silicates	1.84	2.24	1.56	2.20	2.22	2.78	2.20	2.76	2.24	1.98	2.26	2.98	2.27
Organic	1.5	0.8	1.3	1.5	0.8	1.2	1.5	1.6	0.6	0.9	0.6	1.6	1.18
matter													
COD	20	22	24	18	24	18	18	16	18	20	20	18	19.66
Nitrites	0.03	0.03	0.02	0.04	0.04	0.06	0.06	0.04	0.04	0.04	0.02	0.02	0.03

Station-II Table 2-Monthly variation of Physico-chemical parameters of Peddacheruvu Lake water

STATION-II

Table 2-Monthly variation of Physico-chemical parameters of Peddacheruvu Lake water





STATION-III

Table 1-Monthly variation of Physico-chemical parameters of Peddacheruvu Lake water

	Oct-	Nov	Dec	Jan-	Feb	Mar	Apr	May	June	July	Aug	Sep-	Avg
	2018			2019								19	
Temp (in ⁰ C)	22.4	22.1	22.0	23.5	24.6	24.9	25.2	25.6	25.9	24.2	23.2	22.9	23.87
pH	7.9	7.9	7.8	7.8	7.8	7.9	8.0	8.0	8.0	8.1	8.1	8.2	7.95
Carbonates	41.20	43.20	30.28	48.66	32.44	48.32	40.28	52.22	28.36	38.22	48.96	48.98	41.76
Bicarbonates	234.24	230.22	296.64	239.64	241.72	230.64	250.22	280.22	260.26	261.28	262.64	240.22	252.32
DO	9.6	10.4	8.8	8.8	9.6	9.4	9.4	9.6	10.4	10.6	11.2	10.9	9.89
BOD	10	8	8	10	10	8	6	8	10	10	8	6	8.5
ТН	178	199	220	225	260	192	198	202	194	192	156	220	203
Calcium	45.16	53.22	53.96	51.88	40.22	44.66	47.88	51.24	44.38	43.98	49.28	50.22	48
Magnesium	30.22	29.66	29.98	35.86	31.66	33.22	35.12	39.98	30.22	38.26	30.78	40.22	33.76
TDS	380	360	340	325	375	420	435	410	340	380	310	365	370
Chlorides	160.22	141.26	178.66	132.20	171.26	141.98	160.88	148.88	196.98	161.12	149.22	144.98	157.30

D. Srinivas et al Int J Sci Res Sci & Technol. January-February-2021; 8 (1): 212-222

Sulphates	32	34	34	36	32	32	38	38	36	34	30	28	33.66
Phosphates	0.52	0.68	0.76	0.58	0.58	0.46	0.88	0.48	0.54	0.40	0.64	0.66	0.59
Nitrates	0.64	0.70	0.88	0.86	0.84	0.66	0.62	0.78	1.02	1.08	1.04	1.02	0.84
Silicates	1.99	2.50	2.02	2.99	2.66	2.98	2.40	2.88	2.24	2.02	2.28	2.98	2.49
Organic	1.4	1.0	0.8	1.2	1.4	0.8	1.2	1.4	1.2	0.8	1.2	1.6	1.16
matter													
COD	22	20	24	22	18	22	24	18	16	18	20	20	20.33
Nitrites	0.02	0.02	0.04	0.03	0.04	0.02	0.02	0.04	0.06	0.04	0.02	0.04	0.03

Monthly variation of Physico-chemical parameters of Peddacheruvu Lake water





IV. DISCUSSION AND CONCLUSION

The Analysis of Physico-chemical parameters had indicated the wider human activity and influx of domestic waste into the lake caused to eutrification. The values of physico-chemical parameters assessed are found to be slightly above the permissible limit prescribed by WHO for drinking water. The physicochemical parameters show seasonal fluctuations. The water of present Peddacheruvu lake is useful for drinking, irrigation as well as fish culture.

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